

Chemical Kinetics K J Laidler

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Chemical Kinetics and Inorganic Reaction Mechanisms
Introduction to Chemical Engineering Kinetics and Reactor Design
Chemical Kinetics and Reaction Dynamics

Chemical Kinetics and Catalysis

Selected Readings in Chemical Kinetics

"All fields of chemistry involve the principles of chemical kinetics. Important reactions take place in gases, solutions, and solids. This book provides the necessary tools for studying and understanding interactions in all of these phases. Derivations are presented in detail to make them intelligible to readers whose background in mathematics is not extensive."--BOOK JACKET.

Physical Chemistry

Kinetics of Chemical Processes details the concepts associated with the kinetic study of the chemical processes. The book is comprised of 10 chapters that present information relevant to applied research. The text first covers the elementary chemical kinetics of elementary steps, and then proceeds to discussing catalysis. The next chapter tackles simplified kinetics of sequences at the steady state. Chapter 5 deals with coupled sequences in reaction networks, while Chapter 6 talks about autocatalysis and inhibition. The seventh chapter describes the irreducible transport phenomena in chemical kinetics. The next two chapters

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discuss the correlations in homogenous kinetics and heterogeneous catalysis, respectively. The last chapter covers the analysis of reaction networks. The book will be of great use to students, researchers, and practitioners of scientific disciplines that deal with chemical reaction, particularly chemistry and chemical engineering.

Chemical Education: Towards Research-based Practice

Modeling of Chemical Reactions covers detailed chemical kinetics models for chemical reactions. Including a comprehensive treatment of pressure dependent reactions, which are frequently not incorporated into detailed chemical kinetic models, and the use of modern computational quantum chemistry, which has recently become an extraordinarily useful component of the reaction kinetics toolkit. It is intended both for those who need to model complex chemical reaction processes but have little background in the area, and those who are already have experience and would benefit from having a wide range of useful material gathered in one volume. The range of subject matter is wider than that found in many previous treatments of this subject. The technical level of the material is also quite wide, so that non-experts can gain a grasp of fundamentals, and experts also can find the book useful. A solid introduction to kinetics Material on computational quantum chemistry, an important new area for kinetics Contains a chapter on construction of mechanisms, an approach only found in this book

Fundamental Chemical Kinetics

Chemical Kinetics of Gas Reactions

Chemical Kinetics and Catalysis is a comprehensive guide to chemical kinetics and catalysis, and focuses on the use of computational tools for studying chemical kinetics and catalytic phenomena. Provides a thorough and up-to-date treatment of chemical kinetics and catalysis, combining traditional background information with the latest computational methods for fitting data to appropriate rate equations. Demonstrates how the vastly improved computational tools now available allow application of kinetic concepts to understanding and predicting the behavior of diverse and complex phenomena, including biological systems, semiconductor growth, and corrosion. Contains chapters reviewing of kinetic concepts, introducing kinetics via rate equations and mechanisms, explaining the theory of reaction rates (a section on trajectory calculations to simulate reactions), predicting potential energy surfaces (methods for directing the reaction rate), and discussing catalysis with a focus on modifying the reaction rate. A useful reference guide, providing the essential basics along with numerous solved examples, problems, and illustrative computer programs.

Chemical Kinetics

Discusses chemical reactions, examining the bonding in molecules, how molecules interact, what determines whether an interaction is favourable or not, and what the outcome will be.

Reaction Kinetics: Homogeneous gas reactions

Chemical Kinetics bridges the gap between beginner and specialist with a path that leads the reader from the phenomenological approach to the rates of chemical reactions to the state-of-the-art calculation of the rate constants of the most prevalent reactions: atom transfers, catalysis, proton transfers, substitution reactions, energy transfers and electron transfers. For the beginner provides the basics: the simplest concepts, the fundamental experiments, and the underlying theories. For the specialist shows where sophisticated experimental and theoretical methods combine to offer a panorama of time-dependent molecular phenomena connected by a new rational. Chemical Kinetics goes far beyond the qualitative description: with the guidance of theory, the path becomes a reaction path that can actually be inspected and calculated. But Chemical Kinetics is more about structure and reactivity than numbers and calculations. A great emphasis in the clarity of the concepts is achieved by illustrating all the theories and mechanisms

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with recent examples, some of them described with sufficient detail and simplicity to be used in general chemistry and lab courses. * Looking at atoms and molecules, and how molecular structures change with time. * Providing practical examples and detailed theoretical calculations * Of special interest to Industrial Chemistry and Biochemistry

Deterministic Kinetics in Chemistry and Systems Biology

Molecular Reaction Dynamics

Chemical Kinetics and Transport

Chemical education is essential to everybody because it deals with ideas that play major roles in personal, social, and economic decisions. This book is based on three principles: that all aspects of chemical education should be associated with research; that the development of opportunities for chemical education should be both a continuous process and be linked to research; and that the professional development of all those associated with chemical education should make extensive and diverse use of that research. It is intended for: pre-service and

practising chemistry teachers and lecturers; chemistry teacher educators; chemical education researchers; the designers and managers of formal chemical curricula; informal chemical educators; authors of textbooks and curriculum support materials; practising chemists and chemical technologists. It addresses: the relation between chemistry and chemical education; curricula for chemical education; teaching and learning about chemical compounds and chemical change; the development of teachers; the development of chemical education as a field of enquiry. This is mainly done in respect of the full range of formal education contexts (schools, universities, vocational colleges) but also in respect of informal education contexts (books, science centres and museums).

Introduction to Chemical Kinetics

The first IUPAC Manual of Symbols and Terminology for Physicochemical Quantities and Units (the Green Book) of which this is the direct successor, was published in 1969, with the object of 'securing clarity and precision, and wider agreement in the use of symbols, by chemists in different countries, among physicists, chemists and engineers, and by editors of scientific journals'. Subsequent revisions have taken account of many developments in the field, culminating in the major extension and revision represented by the 1988 edition under the simplified title Quantities, Units and Symbols in Physical Chemistry. This 2007, Third Edition, is a further revision of the material which reflects the experience of the contributors with the previous

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editions. The book has been systematically brought up to date and new sections have been added. It strives to improve the exchange of scientific information among the readers in different disciplines and across different nations. In a rapidly expanding volume of scientific literature where each discipline has a tendency to retreat into its own jargon this book attempts to provide a readable compilation of widely used terms and symbols from many sources together with brief understandable definitions. This is the definitive guide for scientists and organizations working across a multitude of disciplines requiring internationally approved nomenclature.

Kinetic Processes in Gases and Plasmas

Chemical Kinetics and Process Dynamics in Aquatic Systems is devoted to chemical reactions and biogeochemical processes in aquatic systems. The book provides a thorough analysis of the principles, mathematics, and analytical tools used in chemical, microbial, and reactor kinetics. It also presents a comprehensive, up-to-date description of the kinetics of important chemical processes in aquatic environments. Aquatic photochemistry and correlation methods (e.g., LFERs and QSARs) to predict process rates are covered. Numerous examples are included, and each chapter has a detailed bibliography and problems sets. The book will be an excellent text/reference for professionals and students in such fields as aquatic chemistry, limnology, aqueous geochemistry, microbial ecology, marine science,

environmental and water resources engineering, and geochemistry.

Reaction Kinetics

This text presents a balanced presentation of the macroscopic view of empirical kinetics and the microscopic molecular viewpoint of chemical dynamics. This second edition includes the latest information, as well as new topics such as heterogeneous reactions in atmospheric chemistry, reactant product imaging, and molecular dynamics of $H + H_2$.

Modern Physical Chemistry

Selected Readings in Chemical Kinetics covers excerpts from 12 papers in the field of general and gas-phase kinetics. The book discusses papers on the laws of connexion between the conditions of a chemical change and its amount; on the reaction velocity of the inversion of the cane sugar by acids; and the calculation in absolute measure of velocity constants and equilibrium constants in gaseous systems. The text then tackles papers on simple gas reactions; on the absolute rate of reactions in condensed phases; on the radiation theory of chemical action; and on the theory of unimolecular reactions. Papers on the theories of unimolecular reactions at low pressures; on the reaction between hydrogen and

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bromine; and on the oxidation of phosphorus vapor at low pressures are also considered. The book further describes papers on the thermal decomposition of organic compounds from the standpoint of free radicals; as well as on a single chain mechanism for the thermal decomposition of hydrocarbons. The book will be invaluable to students of chemical kinetics.

Reaction Kinetics

Chemical Kinetics and Process Dynamics in Aquatic Systems

The range of courses requiring a good basic understanding of chemical kinetics is extensive, ranging from chemical engineers and pharmacists to biochemists and providing the fundamentals in chemistry. Due to the wide reaching nature of the subject readers often struggle to find a book which provides in-depth, comprehensive information without focusing on one specific subject too heavily. Here Dr Margaret Wright provides an essential introduction to the subject guiding the reader through the basics but then going on to provide a reference which professionals will continue to dip in to through their careers. Through extensive worked examples, Dr Wright, presents the theories as to why and how reactions occur, before examining the physical and chemical requirements for a reaction and

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the factors which can influence these. * Carefully structured, each chapter includes learning objectives, summary sections and problems. * Includes numerous applications to show relevance of kinetics and also provides plenty of worked examples integrated throughout the text.

Kinetics of Chemical Processes

"A pedagogical gem. Professor Readey replaces 'black-box' explanations with detailed, insightful derivations. A wealth of practical application examples and exercise problems complement the exhaustive coverage of kinetics for all material classes." --Prof. Rainer Hebert, University of Connecticut "Prof. Readey gives a grand tour of the kinetics of materials suitable for experimentalists and modellers. In an easy-to-read and entertaining style, this book leads the reader to fundamental, model-based understanding of kinetic processes critical to development, fabrication and application of commercially-important soft (polymers, biomaterials), hard (ceramics, metals) and composite materials. It is a must-have for anyone who really wants to understand how to make materials and how they will behave in service." --Prof. Bill Lee, Imperial College London, Fellow of the Royal Academy of Engineering "A much needed text filling the gap between an introductory course in materials science and advanced materials-specific kinetics courses. Ideal for the undergraduate interested in an in-depth study of kinetics in materials." --Prof. Mark E. Eberhart, Colorado School of Mines This book provides an

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in-depth introduction to the most important kinetic concepts in materials science, engineering, and processing. All types of materials are addressed, including metals, ceramics, polymers, electronic materials, biomaterials, and composites. The expert author with decades of teaching and practical experience gives a lively and accessible overview, explaining the principles that determine how long it takes to change material properties and make new and better materials. The chapters cover a broad range of topics extending from the heat treatment of steels, the processing of silicon integrated microchips, and the production of cement, to the movement of drugs through the human body. The author explicitly avoids "black box" equations, providing derivations with clear explanations.

An Introduction to Chemical Kinetics

Chemical Kinetics

Reaction Kinetics, Volume II: Reactions in Solution deals with the kinetics of reactions in solution and discusses the basic principles and theories of kinetics, including a brief description of homogeneous gas reactions. This book is divided into two chapters. The first chapter focuses on the general principles of reactions in solution that includes reactions between ions and involving dipoles; influence of

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pressure on rates in solution; substituent effects; and homogeneous catalysis in solution. Chapter 2 primarily deals with general features of reactions in solution, emphasizing the relationship between the results of a kinetic investigation and actual reaction mechanism. This volume is intended for undergraduate students of chemistry who have not previously studied chemical kinetics. This book is also useful to more advanced students in other fields, such as biology and physics, who wish to have a general knowledge of the subject.

Kinetics in Materials Science and Engineering

It is sometimes said that the year of birth of physical chemistry was 1887. In that year the journal *Zeitschrift für physikalische Chemie* - the first journal devoted exclusively to physical chemistry - was launched and in its first year published important papers by Arrhenius and van't Hoff. However, a good deal of physical chemistry had been done previously. Two centuries earlier Robert Boyle had been carrying out physico-chemical investigations, and a good case can be made for regarding him as the first physical chemist. His approach to chemistry had a great influence on others, including Isaac Newton. In the eighteenth century Joseph Black and Antoine Lavoisier also did much that can be classed as physical chemistry. In the nineteenth century Robert Bunsen, Michael Faraday, and many others were also contributing to the development of the subject. In this book Professor Laidler gives an account of the scientific development of physical chemistry over the

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years. He begins by discussing just what physical chemistry is, and how it relates to other sciences. He considers some of the difficulties faced by early investigators, as a result of attitudes of the Churches, governments, and even the universities which at first were mainly interested in classical studies. Some account is also given of the way in which physical scientists have communicated with each other. Classical mechanics, and the modifications that had to be made to it, are briefly considered. The bulk of the book is concerned with the main branches of physical chemistry - thermodynamics, kinetic theory, statistical mechanics, spectroscopy, electrochemistry, kinetics, colloid and surface chemistry, and quantum chemistry - and how these subjects have developed up to the present time.

Why Chemical Reactions Happen

Kinetic Processes in Gases and Plasmas provides a survey of studies on transport and chemical kinetic processes in high temperature gases and plasmas. The book is concerned with conditions produced by the interaction of an object with the atmosphere at hypersonic velocities. The text also provides a foundation for the flow field equations which include chemical reactions and other transport processes, and to present in some detail the microscopic considerations underlying these calculations. Chapters are devoted to the discussion of topics such as the molecular theory of transport equations; transport processes in ionized gases; and inelastic energy transfer processes and chemical kinetics. Aerospace engineers,

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physicists, chemists, and astrophysicists will find the book a good reference material.

Quantities, Units and Symbols in Physical Chemistry

Theories of Chemical Reaction Rates

The Second Edition features new problems that engage readers in contemporary reactor design. Highly praised by instructors, students, and chemical engineers, *Introduction to Chemical Engineering Kinetics & Reactor Design* has been extensively revised and updated in this Second Edition. The text continues to offer a solid background in chemical reaction kinetics as well as in material and energy balances, preparing readers with the foundation necessary for success in the design of chemical reactors. Moreover, it reflects not only the basic engineering science, but also the mathematical tools used by today's engineers to solve problems associated with the design of chemical reactors. *Introduction to Chemical Engineering Kinetics & Reactor Design* enables readers to progressively build their knowledge and skills by applying the laws of conservation of mass and energy to increasingly more difficult challenges in reactor design. The first one-third of the text emphasizes general principles of chemical reaction kinetics, setting the stage

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for the subsequent treatment of reactors intended to carry out homogeneous reactions, heterogeneous catalytic reactions, and biochemical transformations. Topics include: Thermodynamics of chemical reactions Determination of reaction rate expressions Elements of heterogeneous catalysis Basic concepts in reactor design and ideal reactor models Temperature and energy effects in chemical reactors Basic and applied aspects of biochemical transformations and bioreactors About 70% of the problems in this Second Edition are new. These problems, frequently based on articles culled from the research literature, help readers develop a solid understanding of the material. Many of these new problems also offer readers opportunities to use current software applications such as Mathcad and MATLAB®. By enabling readers to progressively build and apply their knowledge, the Second Edition of Introduction to Chemical Engineering Kinetics & Reactor Design remains a premier text for students in chemical engineering and a valuable resource for practicing engineers.

The Chemical Kinetics of Enzyme Action

Geochemical kinetics as a topic is now of importance to a wide range of geochemists in academia, industry, and government, and all geochemists need a rudimentary knowledge of the field. This book summarizes the fundamentals of geochemical kinetics with examples drawn especially from mineral dissolution and precipitation. It also encompasses discussion of high temperature processes and

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global geochemical cycle modeling. Analysis of textures of rocks, sediments, and mineral surfaces are incorporated throughout and provide a sub-theme of the book.

Chemical Kinetics

This is the definitive treatise on the fundamentals of statistical mechanics. A concise exposition of classical statistical mechanics is followed by a thorough elucidation of quantum statistical mechanics: postulates, theorems, statistical ensembles, changes in quantum mechanical systems with time, and more. The final two chapters discuss applications of statistical mechanics to thermodynamic behavior. 1930 edition.

Introduction to Chemical Kinetics

This book began as a program of self-education. While teaching under graduate physical chemistry, I became progressively more dissatisfied with my approach to chemical kinetics. The solution to my problem was to write a detailed set of lecture notes which covered more material, in greater depth, than could be presented in undergraduate physical chemistry. These notes are the foundation upon which this book is built. My background led me to view chemical kinetics as closely related to

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transport phenomena. While the relationship of these topics is well known, it is often ignored, except for brief discussions of irreversible thermodynamics. In fact, the physics underlying such apparently dissimilar processes as reaction and energy transfer is not so very different. The intermolecular potential is to transport what the potential-energy surface is to reactivity. Instead of beginning the sections devoted to chemical kinetics with a discussion of various theories, I have chosen to treat phenomenology and mechanism first. In this way the essential unity of kinetic arguments, whether applied to gas-phase or solution-phase reaction, can be emphasized. Theories of rate constants and of chemical dynamics are treated last, so that their strengths and weaknesses may be more clearly highlighted. The book is designed for students in their senior year or first year of graduate school. A year of undergraduate physical chemistry is essential preparation. While further exposure to chemical thermodynamics, statistical thermodynamics, or molecular spectroscopy is an asset, it is not necessary.

Kinetics of Water-Rock Interaction

The Principles of Statistical Mechanics

The book is a short primer on chemical reaction rates based on a six-lecture first-

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year undergraduate course taught by the author at the University of Oxford. The book explores the various factors that determine how fast or slowly a chemical reaction proceeds and describes a variety of experimental methods for measuring reaction rates. The link between the reaction rate and the sequence of steps that makes up the reaction mechanism is also investigated. Chemical reaction rates is a core topic in all undergraduate chemistry courses.

Modeling of Chemical Reactions

The unusual approach of this text gives final honours and post-graduate students a clear and explanatory account of one of the “harder areas of physical chemistry. The author takes care to provide detailed verbal clarification of the concepts and their importance together with full explanations of the mathematical developments. Her explanations are an essential and vital feature of the text, which is scholarly, lucid and well-written with a combination of depth of coverage and clarity which helps students to work through on their own. A clear and explanatory account of one of the more difficult areas of physical chemistry Provides detailed verbal clarification of the concepts and their importance together with full explanations of the mathematical developments Discusses energy transfer, molecular beam studies of reactive scattering and historical developments and modern kinetics, among other topics

Chemical Kinetics

This book gives a concise overview of the mathematical foundations of kinetics used in chemistry and systems biology. The analytical and numerical methods used to solve complex rate equations with the widely used deterministic approach will be described, with primary focus on practical aspects important in designing experimental studies and the evaluation of data. The introduction of personal computers transformed scientific attitudes in the last two decades considerably as computational power ceased to be a limiting factor. Despite this improvement, certain time-honored approximations in solving rate equations such as the pre-equilibrium or the steady-state approach are still valid and necessary as they concern the information content of measured kinetic traces. The book shows the role of these approximations in modern kinetics and will also describe some common misconceptions in this field.

Chemical Kinetics

In this new textbook on physical chemistry, fundamentals are introduced simply yet in more depth than is common. Topics are arranged in a progressive pattern, with simpler theory early and more complicated theory later. General principles are induced from key experimental results. Some mathematical background is

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supplied where it would be helpful. Each chapter includes worked-out examples and numerous references. Extensive problems, review, and discussion questions are included for each chapter. More detail than is common is devoted to the nature of work and heat and how they differ. Introductory Caratheodory theory and the standard integrating factor for dG_{rev} are carefully developed. The fundamental role played by uncertainty and symmetry in quantum mechanics is emphasized. In chemical kinetics, various methods for determined rate laws are presented. The key mechanisms are detailed. Considerable statistical mechanics and reaction rate theory are then surveyed. Professor Duffey has given us a most readable, easily followed text in physical chemistry.

Kinetics of Chemical Reactions

The serious study of the reaction mechanisms of transition metal complexes began some five decades ago. Work was initiated in the United States and Great Britain; the pioneers of that era were, in alphabetical order, F. Basolo, R. E. Connick, I. O. Edwards, C. S. Garner, G. P. Haight, W. C. E. Higginson, E. I. King, R. G. Pearson, H. Taube, M. I. Tobe, and R. G. Wilkins. A larger community of research scientists then entered the field, many of them students of those just mentioned. Interest spread elsewhere as well, principally to Asia, Canada, and Europe. Before long, the results of individual studies were being consolidated into models, many of which traced their origins to the better-established field of mechanistic organic

chemistry. For a time this sufficed, but major revisions and new assignments of mechanism became necessary for both ligand substitution and oxidation-reduction reactions. Mechanistic inorganic chemistry thus took on a shape of its own. This process has brought us to the present time. Interests have expanded both to include new and more complex species (e.g., metalloproteins) and a wealth of new experimental techniques that have developed mechanisms in ever-finer detail. This is the story the author tells, and in so doing he weaves in the identities of the investigators with the story he has to tell. This makes an enjoyable as well as informative reading.

Chemical Kinetics and Dynamics

The World of Physical Chemistry

This second, extended and updated edition presents the current state of kinetics of chemical reactions, combining basic knowledge with results recently obtained at the frontier of science. Special attention is paid to the problem of the chemical reaction complexity with theoretical and methodological concepts illustrated throughout by numerous examples taken from heterogeneous catalysis combustion and enzyme processes. Of great interest to graduate students in both

chemistry and chemical engineering.

The Chemical Kinetics of Excited States

Chemical Kinetics bridges the gap between beginner and specialist with a path that leads the reader from the phenomenological approach to the rates of chemical reactions to the state-of-the-art calculation of the rate constants of the most prevalent reactions: atom transfers, catalysis, proton transfers, substitution reactions, energy transfers and electron transfers. For the beginner provides the basics: the simplest concepts, the fundamental experiments, and the underlying theories. For the specialist shows where sophisticated experimental and theoretical methods combine to offer a panorama of time-dependent molecular phenomena connected by a new rational. Chemical Kinetics goes far beyond the qualitative description: with the guidance of theory, the path becomes a reaction path that can actually be inspected and calculated. But Chemical Kinetics is more about structure and reactivity than numbers and calculations. A great emphasis in the clarity of the concepts is achieved by illustrating all the theories and mechanisms with recent examples, some of them described with sufficient detail and simplicity to be used in general chemistry and lab courses. * Looking at atoms and molecules, and how molecular structures change with time. * Providing practical examples and detailed theoretical calculations * Of special interest to Industrial Chemistry and Biochemistry

Principles of Chemical Kinetics

Molecular reaction dynamics is the study of chemical and physical transformations of matter at the molecular level. The understanding of how chemical reactions occur and how to control them is fundamental to chemists and interdisciplinary areas such as materials and nanoscience, rational drug design, environmental and astrochemistry. This book provides a thorough foundation to this area. The first half is introductory, detailing experimental techniques for initiating and probing reaction dynamics and the essential insights that have been gained. The second part explores key areas including photoselective chemistry, stereochemistry, chemical reactions in real time and chemical reaction dynamics in solutions and interfaces. Typical of the new challenges are molecular machines, enzyme action and molecular control. With problem sets included, this book is suitable for advanced undergraduate and graduate students, as well as being supplementary to chemical kinetics, physical chemistry, biophysics and materials science courses, and as a primer for practising scientists.

Chemical Kinetics and Inorganic Reaction Mechanisms

This text teaches the principles underlying modern chemical kinetics in a clear, direct fashion, using several examples to enhance basic understanding. Solutions

to selected problems. 2001 edition. /div

Introduction to Chemical Engineering Kinetics and Reactor Design

Chemical Kinetics of Gas Reactions explores the advances in gas kinetics and thermal, photochemical, electrical discharge, and radiation chemical reactions. This book is composed of 10 chapters, and begins with the presentation of general kinetic rules for simple and complex chemical reactions. The next chapters deal with the experimental methods for evaluating chemical reaction mechanisms and some theories of elementary chemical processes. These topics are followed by discussions on certain class of chemical reactions, including unimolecular, bimolecular, and termolecular reactions. The remaining chapters examine gas reactions, such as molecular collisions, photochemical reactions, chemical reactions in electrical discharge, chain reactions, and combustion. This book will be of value to reaction kinetics engineers and researchers.

Chemical Kinetics and Reaction Dynamics

The range of courses requiring a good basic understanding of chemical kinetics is extensive, ranging from chemical engineers and pharmacists to biochemists and

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providing the fundamentals in chemistry. Due to the wide reaching nature of the subject readers often struggle to find a book which provides in-depth, comprehensive information without focusing on one specific subject too heavily. Here Dr Margaret Wright provides an essential introduction to the subject guiding the reader through the basics but then going on to provide a reference which professionals will continue to dip in to through their careers. Through extensive worked examples, Dr Wright, presents the theories as to why and how reactions occur, before examining the physical and chemical requirements for a reaction and the factors which can influence these. * Carefully structured, each chapter includes learning objectives, summary sections and problems. * Includes numerous applications to show relevance of kinetics and also provides plenty of worked examples integrated throughout the text.

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